

# Close Encounters of Three Reaction Kinds

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#### Three Growing Areas:

Photochemistry



Flow Chemistry



Electrochemistry













#### Photochemistry

- Uses light as a catalyst.
- Alters chemical energy states.
- Lower reaction requirements than traditional pathways.













### Photochemistry

Important for drug-development & organic synthesis.

- High atom efficiency.
- High energy efficiency.
- Beneficial for stereoselectivity.

Challenges have been:

- Reproducibility.
- Scale-up efficiency.











# "Homemade" Photochemistry

"Homemade" photoreactors -

- Unreproducible results
- Potentially dangerous

Important surface area / volume ratio

- decreasing reaction efficiency with scale.













#### Flow Chemistry

- Alternative to batch chemical production.
- Enables continuous reactions.
- Improved heat transfer via increased surface area / volume ratio.
- Integrates with reaction monitoring systems.











#### Flow Chemistry

Important for industrial scale production.

- Safe processing of highly reactive reagents.
- · Simulates large-scale production on a smaller scale.
- Reduce energy and reactant waste.

Challenges have been:

- Heterogeneous reactions
- Accessibility.





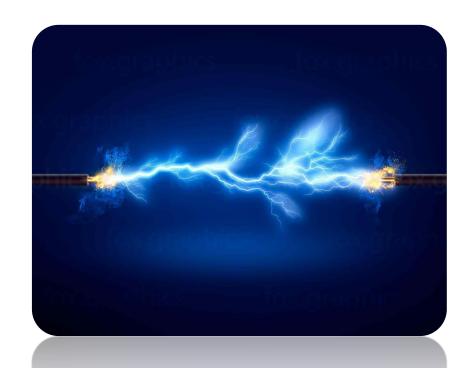






#### Electrochemistry

- Uses electrons as a reactant.
- Enables redox reactions with fewer chemical reagents.
- More atom efficient.
- Highly selective reaction pathways.













#### Electrochemistry

Important for organic chemistry:

- Reduces chemical waste.
- Selective synthesis of novel compounds.

Challenges have been:

- High sensitivity to gases.
- Purity requirements of electrodes.
- Reproducibility.













### "Homemade" Electrochemistry

"Homemade" Electrochemistry reactors -

- Inconsistent distances between electrodes.
- This risks influencing the voltage & reaction success/efficiency.
- Hazard of electrodes touching in flammable solvent.













### Historically...

Photochemistry, Flowchemistry, and Electrochemistry in the past were relatively niche areas of chemistry.

Whilst popular, growth in these areas has been hindered by:

- Technical limitations
- High barriers to entry
- Lack of established data













## Technology - Photochemistry

- Improved, accessible LED technology
- Moving away from Hg lamps which emit a wide range of wavelengths.
- More selective reactions, less energy waste, and less risk from UV.
- Improved methods for batch across varied scales.













#### Technology – Flow Chemistry

Development of standard flow reactors for varied requirements, including solid-liquid / solid-liquid-gas reactions.

- More accessible in R&D
- PFRs (Plug Flow Reactors), CSTRs (Continuous Stirred Tank Reactors), PBRs (Packed Bed Reactors), BCRs (Bubble Column Reactors).













#### Flow Chemistry + Photochemistry

- Flow chemistry is optimal for scaling photochemistry.
- Modular / specialist reactors to accommodate high requirements or changing needs.













### Technology

#### Flow Chemistry

- Automation real-time monitoring and programmable control.
- Modular design flexible reactor usage for optimising processes.
- Reporting Greater accessibility to experimental data.













## Technology

#### Electrochemistry

- Improvements in electrode mounting design for consistent reporting.
- Control of inert atmospheres, stirring and heating in parallel
- Potentiostats for cyclic voltammetry are more powerful and come with improved reporting/data processing.













#### Impact on the Chemistry

- These new developments in chemistry equipment enable greater flexibility and control.
- By simplifying equipment, chemists can focus on the chemistry.
- Photo-, electro- and flow chemistry are more accessible and have clear methods for scaling up.







